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III.																														9
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2.		가																												9
IV.																														13
V.																														16
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Fig 1. Drawing of enamel specimen · · · · · · · · · · · · · · · · · · ·
Fig 2. Removable oral appliance attached with demineralized enamel specimen
Fig 3. Quantitative mineral volume change(%) of enamel · · · · · · · 10
Fig 4. Average mineral volume change of enamel specimen with
remineralization solution "R"11
Fig 5.Average mineral volume change of enamel specimen with saline · · · · 11
Fig 6. Average mineral volume change of enamel specimen with Senstime®
Table 1. Initial composition of demineralization solution · · · · · · · · · · · · · · · · · · ·
Table 2. Initial composition of remineralization solution · · · · · · · · · · · · · · · · · · ·
Table 3. Experimental solution · · · · · · · · · · · · · · · · · · ·
Table 4. Remineralization peak depth from the surface • • • • • • • • 10

가 가 가 가 "R " microradiography 가 가 30 15 "R " 3 5  $(Senstime^{\mathbb{R}})$ saline, . 15 microradiography 가  $Senstime^{\mathbb{R}}$ "R" Saline 1. .(P < 0.05)"R " 2. .(P < 0.05)"R " 가 가

- vii -

"R" 가

( : )

I.

, 가 .<sup>18)</sup>

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가 가 가 .

가 .<sup>4,7)</sup>

가 . Moreno<sup>19)</sup>

- 1 -

가 pН 가 가 가 dicalcium phosphate dihy drate (DCPD) , Chow<sup>4)</sup> 가 가 가 Ten Cate<sup>26)</sup>  $Koulourdes^{\scriptscriptstyle{12)}}$ 가 가 가 가  $Silverstone^{21)}$ 가 가 가 . Ten Cate<sup>27)</sup> cry stallite lesion 가 Ten Cate<sup>28)</sup> 1ppm 가 가 - 2 -

		2ppm	
	가		,
		가	
가			가
	36,39)	가	
microradiograph			

II.

1.

7 30

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、 7 (Senstime<sup>®</sup>, Ildong Co., Korea)

2. フト curette フト pumice

, 기 2-3mm window , Gold grid

nail vanish .
bonding resin(Scotchbond MP, 3M ,U.S.A) 가

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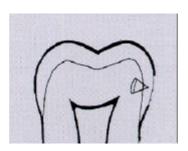


Fig 1. Drawing of enamel specimen

3.

1 )Stock 30% lactic acid (Sigma Co.U.S.A, 90.08, 1.080) 1M (Sigma Co.U.S.A, 147.0) lactic acid 0.3M(Sigma Co., 136.1) 0.9M100 (790PersonalIC.Metrohm.Ion analysis.Switzerland) 2) 9.5mM, NaN3 3.08mM, lactic acid Stock 15mM, 0.155<sup>29)</sup>가 100m M , 8M KOH pH meter(Advanced Ion Selective Meter 920. Orion Research, U.S.A) pH 4.3 100 (790PersonalIC.Metrohm.Ion analysis.Switzerland) 3) 10ppm (100ppm NaF Standard solution, Orion Research Inc., U.S.A.) 2ppm Table 2 stock protocol

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Table 1. Initial composition of demineralization solution

Composition	Concentration (mM)	
Lactic acid(mM)	100	
Calcium (mM)	15.1	
Phosphate(mM)	9.6	
Sodium azide(mM)	3.08	
pН	4.3	
Degree of saturation	0.155	

Table 2. Initial composition of remineralization solution

Composition	Concentration (mM)	
Lactic acid(mM)	10	
Calcium (mM)	31.2	
Phosphate(mM)	12.0	
Sodium azide(mM)	3.08	
Fluoride(ppm)	2	
pН	4.3	
Degree of saturation	0.339	

4.
50cc 30cc
37 . 2 7.
microradigraph .

5.

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Fig 2. Removable oral appliance attached with demineralized enamel specimen

6.

15

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10

Table 3. Experimental solution

Group	Treatment sol	ution	
	Saline Senstime <sup>®</sup>	"R"	
7.			
15		,	,
5mA, 37Kvp	1 30	microradiograph	
microradios	graph	100	Coolpix 950
(Nikon corporation.Japa	an)		Image
analysis program			
1	Excel(MS Office 200	0,Microsoft U.S.A)	
가	가		
Mineral intens	ity가 가		
Kruscal-Wallis test	Duncan		

III.

1.			
Microrac	diography		
	erosion	가 . 30	
		198 ± 95u m	가
2.			
	가		가
		"R" $34.53 \pm 14.08\%$ ,	
	$22.06 \pm 13.67\%$ , Senstime <sup>®</sup>	28.97 ± 9.18% 기	
.(T able 4,	Fig 3)		가
mineral i	intensity가 ,	가	
	$97 \pm 15.08$ um,	$37 \pm 8.72$ um, Senstime <sup>®</sup>	73 <b>±</b>
17.01um	.(T able 4)		

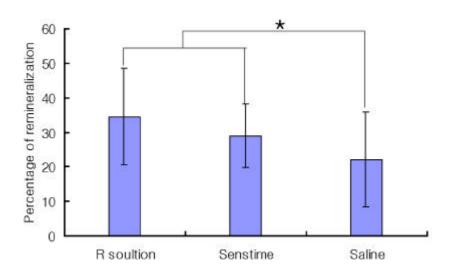
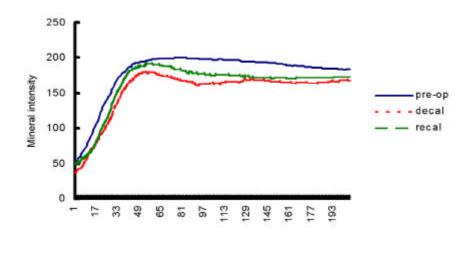


Fig 3. Quantitative mineral volume change(%) of enamel

Table4. Remineralization peak depth from the surface

Group	Remineralization peak depth(µm)
R solution	97 ± 15.08 *
Saline	$37 \pm 8.72$
S en stim e <sup>®</sup>	$73 \pm 17.01$

<sup>\*</sup> significantly different (p<0.05)



 $\mu m$ 

Fig 4. Average mineral volume change of enamel specimen with remineralization solution "R"

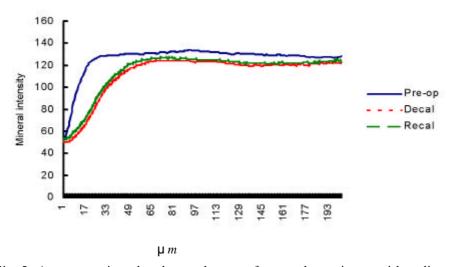


Fig 5. Average mineral volume change of enamel specimen with saline

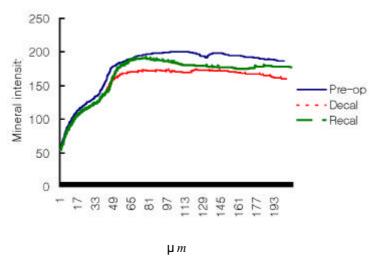


Fig 6. Average mineral volume change of enamel specimen with  $Senstime^{\circledR}$ 

Saline "R" Sensitime 
$$^{\circledR}$$
 
$$(P\!<\!0.05)$$
 
$$(\ P\!<\!0.05).$$

IV.

			30)		가						
pН,	가	가						pH <sup>29,34,35,3</sup>	, <sup>38,40)</sup> <b>가</b>		
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						(supers	saturat	ed)			
							가				
									가		
			(under sa	turated)					31)		
	(Deg	ree	of saturation	n) pH			, ,				
			•								
							D.#	devold <sup>3)</sup>			
			가	•			Druc	uevoia		가	
가			·				I	Bunnocor	re <sup>24)</sup>	•	
					가						
		. N	Ioreno <sup>19)</sup>						,		
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				フ	<b>'</b> }	가					
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			가								가
			71	가				,			<b>~</b> I

DCPD				가	
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7			71		
35)			가	•	
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Ca-P crystal					,
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가			,		
			artf	icial gel	
9,22), partiall	y saturated buffer		15), plaque b	acteria <sup>5)</sup>	
,	partially saturate	ed buffer		gel	
		41)			
		가			
	. pH <sup>38)</sup>				
	•		4.3	41)	38)
	가		-		
	. рН		35)	nЦ	4.0
5.5				pm	7.0
5.5	pН				

- 14 -

pH가 . Margolis 16) 가 1ppm 가 , 1ppm 가 가 가 가 가 0.339 36) 가 가 가 가 23)  $^{1)}$ ,microradiograph $^{10)}$ , microhardnesslight scattering 25) 가 . Microradigraphy gold grid grid window17,33,37) plaque fluid 30)

- 15 -

가 in vitro 20) In vivo pН, 32) 가 가 가 saline  $Senstim\,e^{\tiny{\textcircled{R}}}$ 가 90.48 ppm 가 가 가 가 Lar sen 14) 0.001ppm 1ppm 42) 가 flouroaptite가  $Senstim\,e^{{\textstyle \,\overline{\!\!\!\!\!\!R}}}$ 가 , bicarbonate 가 가 m acrom olecle pellicle

8,11) •

 $Senstime^{\mathbb{R}}$ 

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가 . **V** .

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S en stim e <sup>®</sup>		microradiog	graph	
				٠
1.		Senstime <sup>®</sup>	Saline	
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## **Abstract**

## In vivo Quantitative Analysis of Remineralization effect of remineralization solution "R" of incipient enamel dental caries

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Dental caries is a chronic disease that causes the destruction of tooth structure by the interaction of plaque bacteria, food debris, and saliva.

There has been attempts to induce remineralization by supersaturating the intra-oral environment around the surface enamel, where there is incipient caries.

In this study, supersaturated remineralized solution "R" was applied to specimens with incipient enamel caries, and the quantitative analysis of remineralization was evaluated using microradiography. Thirty subjects volunteered to participate in this study. Removable appliances were constructed for the subjects, and the enamel specimen with incipient caries were embedded in the appliances. The subjects wore the intra-oral appliance for 15 days except while eating and sleeping.

The removable appliance were soaked in supersaturated solution "R", saline, or Senstime<sup>®</sup> to expose the specimen to those solutions three times a day, 5 minutes each time. After 15 days, microradiography was retaken to compare and evaluate remineralization.

The results were as the following:

1. The ratio of remineralized area to demineralized area was significantly higher in the supersaturated solution "R" and Senstime  $^{\circledR}$  than in the saline (p<0.05)

2. Remineralization in the supersaturated buffer solution "R" occurred in the significantly deeper parts of the tooth, compared to the Senstime  $^{\circledR}$  group containing high concentration of fluoride.(p<0.05)

As in the above results, the remineralization effect of remineralized buffer solution "R" on incipient enamel caries has been proven. For clinical utilization, further studies on soft tissue reaction and the effect on dentin and cementum are necessary.

In conclusion compared to commercially available fluoride solution, remineralization solution "R" showed better remineralization effect on early enamel caries lesion, so it is considered as effecient solution for clinical application

Key words: enamel caries remineralization solution, remineralization,

removable oral appliance, microradiography